

$$\frac{X \text{ mmol pdt}}{0.0872} = \frac{\frac{1.06}{2H}}{\frac{3}{3H}}$$

$$X = 0.0462 \text{ mmol} = 31.4 \text{ mg product}$$

$$\frac{(31.4 \text{ mg})}{32.5 \text{ mg}} * 100 = 97 \text{ wt\%}$$

